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Note

Increasing the optical rotation of L-carbidopa by reaction with dicarbonyl compounds

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Summary

To enhance optical rotation, the hydrazino moiety of L-carbidopa was derivatized to a pyrazole ring by reaction with a β -dicarbonyl compound, 2,4-pentanedione, 1,1,1-trifluoro-2,4-pentanedione or malondialdehyde. Reaction with 2,4-pentanedione produced a product with an increased specific rotation in a derivatization completed in 20 min.

L-Carbidopa (*S*-(–)-3-(3,4-dihydroxyphenyl)-2-hydrazino-2-methylpropanoic acid monohydrate) is reported to have a specific rotation of -17.3° in methanol. Both the United States and British Pharmacopeia make use of the chelation of L-carbidopa with aluminum, using aqueous aluminum chloride as the solvent in a polarimetric analysis. This provides a negligible increase in optical rotation. Under these conditions, the USP requirement is a specific rotation, $[\alpha]_D$, between -21 and -23.5° at 25°C , and the BP requires between -22.5 and -26.5° measured at 20°C , the difference being due to the temperature of the measurement. Ceccarini and Maione (1987) proposed measurement at 25°C using a 3.05%

solution in a mixture of equal volumes of concentrated formic and acetic acids containing *p*-dimethylaminobenzaldehyde. Hydrazone formation under these conditions was completed in 10 min. The derivative exhibited a specific rotation of about $+190^\circ$, and observed rotations are linear from 0.6 to 3% (w/v). Although this proposal has merit, we have reservations on the use of the corrosive acids.

Dee (1971) determined hydrazine and methylhydrazine by gas chromatography, using 2,4-pentanedione to make the corresponding pyrazoles. The reaction was completed within 1 h over the pH range of 6–9. Because the catechol function is unstable in this pH range, we studied the reaction of the hydrazino moiety of L-carbidopa with β -dicarbonyl compounds only under acidic conditions (see Fig. 1). Pyrazole derivatization of L-carbidopa was studied at various acid pH values using 2,4-pentanedione, 1,1,1-trifluoro-2,4-pentanedione or

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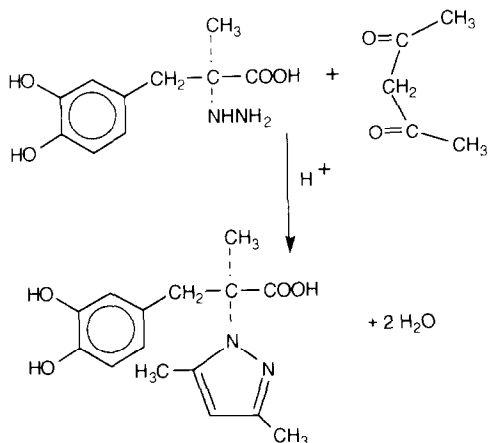


Fig. 1. Reaction of L-carbidopa with 2,4-pentanedione to form a pyrazole with enhanced optical rotation.

the tetraethoxy acetal of malondialdehyde. Solution discoloration was encountered using malondialdehyde in acid, possibly due to aldol reactions. The trifluoro dicarbonyl gave a larger increase in observed rotation; however, its reaction rate was slower than that of 2,4-pentanedione.

Chafetz (1991) proposed changing the wavelength for measurement of optical rotation from the sodium D-line (589 nm) to the lower wavelengths afforded by isolating the lines from a mercury lamp with the use of a photoelectric polarimeter. Such instruments are nearly ubiquitous in pharmaceutical laboratories, their precision is much better than that of visual polarime-

ters, mercury lamps last twice as long as sodium arc lamps and the rotations at 365 nm are generally 3 times greater than at 589 nm.

The recommended procedure is simple: dissolve 40 mg (0.16 mmol) of L-carbidopa in about 9 ml of 1 N hydrochloric acid in a 10 ml volumetric flask. Add 320 μ l (3.1 mmol) of 2,4-pentanedione, bring to the mark with hydrochloric acid and mix well. Allow to stand for 20 min and determine the optical rotation of the solution in a 1 dm polarimeter cell using the 365 nm wavelength. The first-order reaction rate for disappearance of L-carbidopa was about 0.34 min^{-1} at 25°C . The specific rotation, $[\alpha]_{365}$, at 25°C was about -139° , and the relation between observed rotation and concentration was linear from 0.1 to 1.0% (w/v) of L-carbidopa. The derivative exhibiting the increased optical rotation is presumed to be *S*-($-$)-3-(3,4-dihydroxyphenyl)-2-(3,5-dimethylpyrazolo)-2-methylpropanoic acid.

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